

BASIC THERMODYNAMICS

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| Course code | 20ES1304 | Year | II | Semester | I |
| Course category | Engineering Science Course | Branch | ME | Course Type | Theory |
| Credits | 3 | L-T-P | 3-0-0 | Prerequisites | Nil |
| Continuous Internal Evaluation: | 30 | Semester End Evaluation: | 70 | Total Marks: | 100 |

| CO | Statement | Level | Units |
|-----|---------------------------------------------------------------------------------------------------------------------------------------------------|-------|-----------|
| CO1 | Understand the fundamental laws of thermodynamics, concept of reversibility, phase transformation of materials and various thermal cycles. | L2 | 1,2,3,4,5 |
| CO2 | Apply the energy conservation for closed and open cycle systems. | L3 | 1,2 |
| CO3 | Apply the directional law for various cyclic devices named as Heat Engine, Heat Pump and Refrigerator. | L3 | 2,3 |
| CO4 | Analyze availability and entropy of various perfect gas as well as phase transforming thermodynamic processes. | L4 | 3,4 |
| CO5 | Analyze the performance of different thermodynamic cycles. | L4 | 5 |
| CO6 | Analyze the given scenario, use appropriate techniques and write an effective report. | L4 | 2,3,4,5 |

Contribution of Course Outcomes towards achievement of Program Outcomes

| | PO1 | PO2 | PO3 | PO4 | PO5 | PO6 | PO7 | PO8 | PO9 | PO10 | PO11 | PO12 | PSO 1 | PSO 2 |
|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|------|------|------|-------|-------|
| CO1 | 3 | | | | | | | | | | | | 3 | |
| CO2 | 3 | 3 | | | | | | | | | | | 3 | |
| CO3 | 3 | 3 | | | | 2 | | | | | | 2 | 3 | |
| CO4 | 3 | 3 | | | | | | | | | | | 3 | |
| CO5 | 3 | 3 | | | | 2 | | | | | | 2 | 3 | |

Syllabus

| Unit No | Contents | COs |
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| I | INTRODUCTION: Macroscopic and microscopic viewpoints, definitions of thermodynamic terms, quasi – static process, point and path function, forms of energy, ideal gas and real gas, Zeroth law of thermodynamics. FIRST LAW OF THERMODYNAMICS (Closed System): Joule’s experiment - first law of thermodynamics, corollaries- perpetual motion machines of first kind, First law applied to non-flow process. | CO 1, CO 2, |
| II | FIRST LAW OF THERMODYNAMICS (Open System): First law applied to flow process, Steady flow energy equation- limitations of first law of thermodynamics. SECOND LAW OF THERMODYNAMICS: Kelvin - Planck statement and Clausius statement and their equivalence, corollaries - perpetual motion machines of second kind - reversibility and irreversibility, cause of irreversibility | CO 1, CO 2, CO 3, CO 6 |
| III | ENGINEERING DEVICES: Carnot cycle, heat engine, heat pump and refrigerator, Carnot theorem, Carnot efficiency. ENTROPY: Clausius inequality -Concept of Entropy- entropy equation for different processes and systems, Maxwell relations, TdS equations. | CO 1, CO 3, CO 4, CO 6, |

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| IV | <p>AVAILABILITY AND IRREVERSIBILITY: Definition of exergy and energy, expressions for availability and irreversibility. Availability in steady flow, non-flow processes, irreversibility.</p> <p>PROPERTIES OF STEAM AND USE OF STEAM TABLES: Pure Substances, P-V-T surfaces, T-s and h-s diagram, Mollier chart, dryness fraction, property tables, analysis of steam undergoing various thermodynamic processes using Mollier chart– steam calorimetry.</p> | CO 1, CO 4, CO 6, |
| V | <p>GAS POWER CYCLES: Otto, Diesel, Dual Combustion cycles- Description and representation on P–V and T-S diagram, Thermal Efficiency, Mean Effective Pressures on Air standard basis – comparison of Cycles.</p> <p>THERMODYNAMIC CYCLES: Sterling Cycle, Atkinson Cycle, Ericsson Cycle, Lenoir Cycle, Brayton Cycle – Description and representation on P–V and T-S diagram, Thermal Efficiency, Mean Effective Pressures on Air standard basis – comparison of Cycles.</p> | CO 1, CO 5, CO 6, |

| Learning Resource |
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| Text books: |
| <ol style="list-style-type: none"> 1. P.K.Nag, Engineering Thermodynamics, 5/e, Tata McGraw Hill, 2013. 2. Yunus A. Cengel, Michael A. Boles, Thermodynamics, 7/e, Tata McGraw Hill, 2011. |
| Reference books |
| <ol style="list-style-type: none"> 1. J.B.Jones and G.A.Hawkins, Introduction to Thermodynamics, 2/e, John Wiley & Sons, 2012. 2. Moran, Michael J. and Howard N. Shapiro, Fundamentals of Engineering Thermodynamics, 3/e, Wiley, 2015 3. Claus Borgnakke Richard E. Sonntag, Fundamentals of Thermodynamics, 7/e, Wiley, 2009 4. R.K. Rajput, S.Chand& Co., Thermal Engineering, 6/e, Laxmi publications, 2010. |
| e- Resources & other digital material |
| 1. https://nptel.ac.in/courses/112/105/112105266/ |
| 2. https://nptel.ac.in/courses/103/103/103103144/ |
| 3. https://nptel.ac.in/courses/112/105/112105220/ |
| 4. https://nptel.ac.in/courses/101/104/101104067/ |
| 5. https://nptel.ac.in/courses/101/104/101104063/ |
| 6. https://nptel.ac.in/courses/103/104/103104151/ |